

Ions on balanced atom

Name _____

Date _____

Class _____

3-3 Practice Problems

1. How many protons and electrons are present in a vanadium atom?

23 p⁺
23 e⁻

2. How many protons and electrons are present in a nitrogen atom?

7 p⁺
7 e⁻

3. How many protons and electrons are present in an argon atom?

18 p⁺
18 e⁻

4. How many protons and electrons are present in a potassium atom?

19 p⁺
19 e⁻

5. How many protons and electrons are present in a platinum atom?

78 p⁺
78 e⁻

6. What is the name of the element that has atoms that contain 5 protons?

Boron

7. What is the name of the element that has atoms that contain 17 protons?

Chlorine

8. What is the name of the element that has atoms that contain 25 protons?

Manganese

9. What is the name of the element that has atoms that contain 82 protons?

Lead (Pb)
Plumbum

10. What is the name of the element that has atoms that contain 92 protons?

Uranium is used in nuclear power plants.

11. Write the chemical symbol for the ion with 12 protons and 10 electrons.

Mg²⁺ 12-10

12. Write the chemical symbol for the ion with 74 protons and 68 electrons.

W⁶⁺ 74-68
Wolfram

13. Write the chemical symbol for the ion with 95 protons and 89 electrons.

Am⁶⁺ 95-89

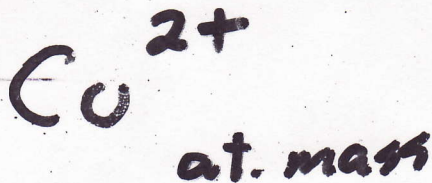
14. Write the chemical symbol for the ion with 33 protons and 36 electrons.

As³⁻ 33-36 = -3

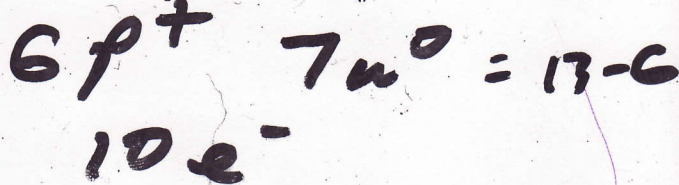
3-3 Practice Problems (continued)

neutrons = atomic mass - atomic

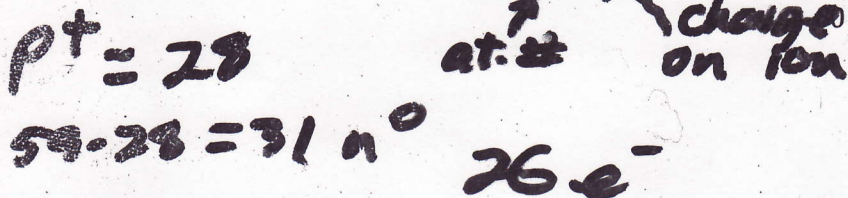
15. Write the chemical symbol for the ion with 29 protons and 27 electrons.



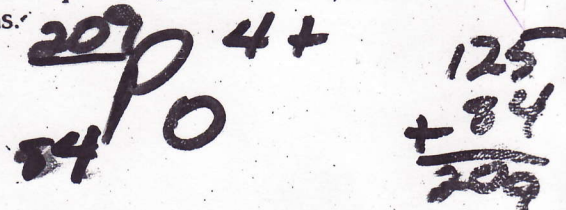
21. How many protons, neutrons, and electrons are present in the $^{13}\text{C}^{4-}$ ion?



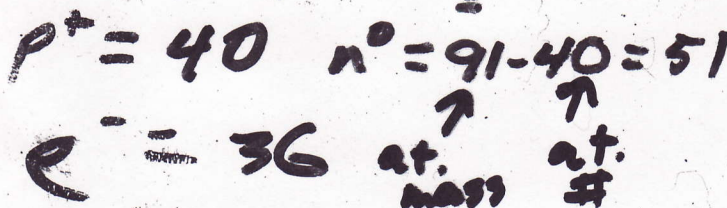
16. How many protons, neutrons, and electrons are present in the $^{59}\text{Ni}^{2+}$ ion?



22. Write the complete chemical symbol for the ion with 84 protons, 125 neutrons, and 80 electrons.



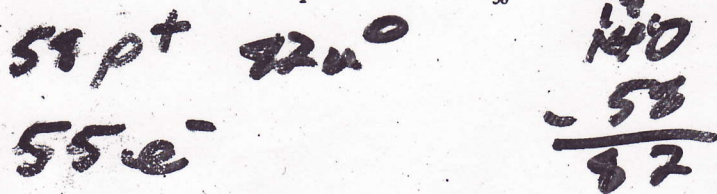
17. How many protons, neutrons, and electrons are present in the $^{91}\text{Zr}^{4+}$ ion?



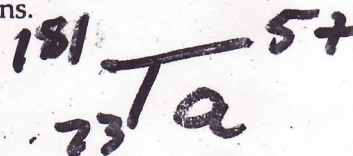
23. Write the complete chemical symbol for the ion with 27 protons, 32 neutrons, and 25 electrons.



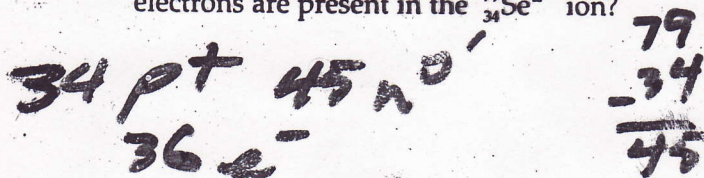
18. How many protons, neutrons, and electrons are present in the $^{140}\text{Ce}^{3+}$ ion?



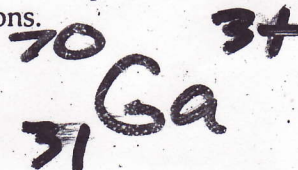
24. Write the complete chemical symbol for the ion with 73 protons, 108 neutrons, and 68 electrons.



19. How many protons, neutrons, and electrons are present in the $^{79}\text{Se}^{2-}$ ion?



25. Write the complete chemical symbol for the ion with 31 protons, 39 neutrons, and 28 electrons.



20. How many protons, neutrons, and electrons are present in the $^{45}\text{Sc}^{3+}$ ion?

